

$$.227 \text{ g H}_2\text{O} \frac{1 \text{ mol H}_2\text{O}}{18.02 \text{ g H}_2\text{O}} = 0.0126 \text{ mol H}_2\text{O}$$

0.0126 moles H₂O

2. Calculate the moles of copper in your sample of copper chloride?

$$0.742 \text{ g Cu} \frac{1 \text{ mol Cu}}{63.546 \text{ g Cu}} = 0.0117 \text{ mol Cu} \quad \underline{0.0117} \text{ moles Cu}$$

3. Calculate the moles of chlorine in your sample of copper chloride?

$$M_{\text{Cl}} = M_{\text{hydrate}} - M_{\text{Cu}} = 0.777 \text{ g} - 0.742 \text{ g} = 0.035 \text{ g Cl} \quad \underline{0.0005} \text{ moles Cl}$$

$$0.035 \text{ g Cl} \frac{1 \text{ mol Cl}}{70.906 \text{ g Cl}} = 0.0005 \text{ mol Cl}$$

4. Write the proper chemical formula for the compound that you tested.

$$\text{Moles Cu} = \frac{.0117}{.0005} = 2.3 \div 3 = 7$$

$$\text{Moles Cl} = \frac{.0005}{.0005} = 1 \div 3 = 3$$

Correct formula for initial hydrated salt _____

5. Explain using chemical equations the process of a) dehydrating the hydrate salt b) dissolving the dehydrated salt c) the change in copper observed after adding the aluminum wire to the solution.

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Gary S. Thorpe



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Confidence Uncertainty Continuum expression This important relationship epitomizes the scientific enterprise as a search for probabilistic rather than absolute truth This two volume set will contain a variety of ways that data quality can be instituted into a science curriculum To support its use many of the examples that I will present involve science teachers as well as student work and feedback from different grade levels and in different scientific disciplines Specific chapters will be devoted to reviewing the academic literature on data quality as well as describing my own personal research on this important but often neglected topic

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