

## SNC2DN -CHEMISTRY Diagnostic Test

Name: \_\_\_\_\_ Date: \_\_\_\_\_

**NOTE:** The purpose of this test is to illustrate your current knowledge about specific domains in science. Your results on this test will in no way directly affect course grade.

**INSTRUCTIONS:** Be as complete as possible. Do not go over the space provided. Fit the most important things that you have to say in the space provided. Please write in complete sentences.

### CHEMISTRY

1. What is matter?
2. What is the difference between an element, an atom, and a compound?
3. What happens in a chemical reaction?
4. Distinguish between a physical and chemical property. Provide examples of each.
5. Label each of the following properties as either physical or chemical:
  - a. Copper sulfate crystals are blue. \_\_\_\_\_
  - b. Gold is an excellent conductor of electricity. \_\_\_\_\_
  - c. Iron rusts when exposed to air and water. \_\_\_\_\_
  - d. Salt is soluble in water. \_\_\_\_\_
  - e. Magnesium reacts with acid to produce hydrogen gas. \_\_\_\_\_
  - f. Gasoline burns in a car engine. \_\_\_\_\_
6. List 5 clues that indicate a chemical change has occurred.
7. (a) What is the formula for density?  
  
(b) Calculate the density of a 500g rectangular block with the dimensions 8cm x 6cm x 5cm.
8. Define the terms: ion, anion and cation.
9. What 3 main categories of elements make up the periodic table?
10. Define the terms periods and families/groups as they relate to the Periodic table.

# Diagnostic Test Chemistry Answer Key

**Barron's Educational Series, Sandra S.  
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**Table of Contents Diagnostic Test Chemistry Answer Key**

1. Understanding the eBook Diagnostic Test Chemistry Answer Key
  - The Rise of Digital Reading Diagnostic Test Chemistry Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Diagnostic Test Chemistry Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Diagnostic Test Chemistry Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Diagnostic Test Chemistry Answer Key
  - Personalized Recommendations
  - Diagnostic Test Chemistry Answer Key User Reviews and Ratings
  - Diagnostic Test Chemistry Answer Key and Bestseller Lists
5. Accessing Diagnostic Test Chemistry Answer Key Free and Paid eBooks
  - Diagnostic Test Chemistry Answer Key Public Domain eBooks
  - Diagnostic Test Chemistry Answer Key eBook Subscription Services
  - Diagnostic Test Chemistry Answer Key Budget-Friendly Options
6. Navigating Diagnostic Test Chemistry Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Diagnostic Test Chemistry Answer Key Compatibility with Devices
  - Diagnostic Test Chemistry Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Diagnostic Test Chemistry Answer Key
  - Highlighting and Note-Taking Diagnostic Test Chemistry Answer Key
  - Interactive Elements Diagnostic Test Chemistry Answer Key
8. Staying Engaged with Diagnostic Test Chemistry Answer Key

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Diagnostic Test Chemistry Answer Key
- 9. Balancing eBooks and Physical Books Diagnostic Test Chemistry Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Diagnostic Test Chemistry Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Diagnostic Test Chemistry Answer Key
  - Setting Reading Goals Diagnostic Test Chemistry Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Diagnostic Test Chemistry Answer Key
  - Fact-Checking eBook Content of Diagnostic Test Chemistry Answer Key
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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