



MASTERY TEST SCIENCE 10

Name: _____
Section: _____

Date: _____

Directions: Read and analyze each item carefully. Write the letter of the correct answer in your answer sheet.

1. How do arteries differ from veins?
 - A. Arteries carry blood.
 - B. Arteries are thin-walled and are used for gas exchange.
 - C. Arteries have thicker wall layers to accommodate the changes in pressure from the heart.
 - D. Arteries have thinner wall layers and valves and move blood by the action of skeletal muscle.
2. Why is the lung considered as the primary organ of the respiratory system?
 - A. Because it is the largest organ of the system.
 - B. Because it has two parts: the right and left lungs.
 - C. Because it is located near the middle of the chest cavity.
 - D. Because it contains the alveoli that is responsible for gas exchange.
3. Smoking is the common cause of emphysema. Which situation will happen when you quit smoking immediately?
 - A. The blood in your lungs will lack oxygen
 - B. The amount of oxygen in your lungs will decrease
 - C. The blood pressure in your pulmonary artery will increase
 - D. The amount of oxygen in your blood stream will, ultimately, increase
4. In cattle, roan coat color (mixed red and white hairs) occurs in the heterozygous (Rr) offspring red (RR) and white (rr) homozygotes. Which cross would produce offspring in the ration of 1 red: 2 roan: 1 white?
 - A. RR x rr
 - B. RR x Rr
 - C. Rr x Rr
 - D. Rr x RR
5. An extra finger in human is rare but is due to a dominant gene. When one parent is normal, and the other parent has an extra finger but is heterozygous for the trait, what is the probability that the first child is normal?
 - A. Zero percent
 - B. Fifty percent
 - C. twenty-five percent
 - D. seventy-five percent
6. Where the electrons are usually found according to the atomic theory?
 - A. In the atomic nucleus.
 - B. Outside the nucleus, most of an atoms' volume is the electron cloud.
 - C. Outside the nucleus, yet near it because they are attracted to protons.
 - D. Anywhere they want to be – no particular location is more likely than any other.
7. What is the name of the model of the atom that scientists use today?
 - A. Billiard Model
 - B. The Nuclear Model
 - C. Plum Pudding Model
 - D. Quantum Mechanical Model
8. Which of the following statements BEST explains the high solubility of sodium chloride (NaCl) in water?
 - A. The ions are free to move.
 - B. There are positive and negative ions present in the substance.
 - C. Sodium chloride break apart into molecules, not an individual atom when dissolve in water.
 - D. The positive and negative ions are attracted to different regions of the polar water molecules.
9. Why do ionic compounds conduct electricity when dissolved in water, but most covalent bonds do not?
 - A. Covalent compounds have lower melting point than ionic compounds.
 - B. Covalent compounds are weakly bonded while ionic compounds are strongly bonded.
 - C. Covalent compounds do not dissolve in water since they are composed of non-polar molecules.

Prepared by:
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