

Honors Chemistry Quiz: Chapter 5 Gas Laws

Name: _____ Period: _____ Date: _____

This quiz is worth 48 points. Only those quizzes completed in black ink will be graded. Useful information: $R = 0.08206 \text{ L-atm/mol-K}$. Good luck!

1) Convert 1.25 atm to mm Hg.

- A) 760 mm Hg
- B) 875 mm Hg
- C) 950 mm Hg
- D) 1000 mm Hg
- E) 1520 mm Hg

2) The volume of a gas is proportional to number of moles of a gas is known as

- A) Avogadro's Law
- B) Ideal Gas Law
- C) Charles's Law
- D) Boyle's Law
- E) Dalton's Law

3) What volume will a balloon occupy at 1.0 atm, if the balloon has a volume of 7.6 L at 3.8 atm?

- A) 2.0 L
- B) 5.0 L
- C) 29 L
- D) 35 L
- E) 17 L

4) To what volume will a sample of gas expand if it is heated from 50.0°C and 2.33 L to 500.0°C ?

- A) 5.58 L
- B) 9.74 L
- C) 10.3 L
- D) 17.9 L
- E) 4.38 L

5) The volume of a gas is proportional to the temperature of a gas is known as

- A) Avogadro's Law
- B) Ideal Gas Law
- C) Charles's Law
- D) Boyle's Law
- E) Dalton's Law

Honors Chemistry Gas Laws Review

United States Air Force Academy

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the United States Air Force Academy text.

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