

Honors Chemistry Worksheet 3

Stoichiometry Practice Problems

Name _____ Period _____ Date _____

Instructions: Balance the following chemical equations and then determine the missing information for each of the conditions given. The four questions related to each equation are independent of one another. Answers for a particular numbered problem should be in the same units as the information given (i.e. grams to grams, moles to moles, particles to particles).

$\text{Al}_2\text{O}_{3(s)} \rightarrow \text{Al}_{(s)} + \text{O}_{2(g)}$ Work Area

1. 5mol _____
2. _____ 2.5mol
3. 25g _____
4. _____ 2.41×10^{24} molecules

$\text{CuO}_{(s)} + \text{H}_2\text{SO}_{4(aq)} \rightarrow \text{CuSO}_{4(aq)} + \text{H}_2\text{O}_{(l)}$

1. _____ .75mol _____
2. 50g _____
3. _____ 3.5mol _____
4. _____ 1.204×10^{23} molecules

$\text{K}_{(s)} + \text{N}_{2(g)} \rightarrow \text{K}_3\text{N}_{(s)}$ Work Area

1. 1.51×10^{23} atoms _____
2. _____ 1.25mol
3. _____ 25 grams

Honors Chemistry Stoichiometry Practice Worksheet

Blue Writing

PT Brinkman



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