

ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পদ্ধতির প্রয়োগের

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

সংকেত, $m = 20$ g

HCl এর আপেক্ষিক ভর,

$$M = 36.5 \text{ g/mol}$$

চলকের আয়তন, $V = 1 \text{ L}$

চলকের ঘনমাত্রা $S = 1$

উদাহরণে দ্রবিত X-পদার্থিতে 10 mg এর মধ্যে কয় পরিমাণ দ্রব
যে X-পদার্থিতে 0.5 g 2% $\left(\frac{W}{V}\right)$ HCl দ্রব দিতে কাজ করা হয়েছে।

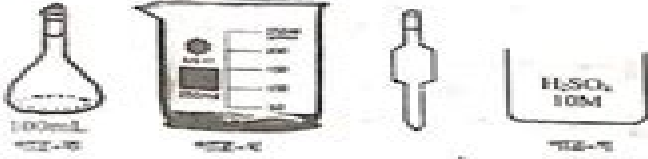
যদিও X-পদার্থটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও
Y-পদার্থটি হলো ফার্মা আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির
মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিম্নে তা বিশ্লেষণ করা
হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি।
এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও
কঠোরভাবে বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05
থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে
কাজ করা হয়। ফলে পরিবেশ দূষণ উল্লেখযোগ্যভাবে কমে আসে।
এছাড়া সেমি-মাইক্রো পদ্ধতিতে H_2S এর পরিবর্তে
কোয়েলিনাইট ব্যবহার করা হয় ফলে পরিবেশ দূষণ ঘটে না।



অতএব, কোনো কোনো ক্ষেত্রে কম কষ্টের রাসায়নিক দ্রব্য ব্যবহৃত হয়।
ফলে পরিবেশ দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি
পরিবেশ বান্দব।

চিত্র-১০



১. টাইটার দ্রবক কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যাঞ্জার প্রকৃতিটি অঙ্কন করে এবং
ব্যাখ্যা করে।
৪. উদাহরণের অংশের দ্রবণ থেকে উপযুক্ত কাচ পাত্র ব্যবহার
করে 1M ঘনমাত্রার দ্রবণ তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪
ও না প্রদেয় উত্তর

৫. বিশ্লেষণের নিমিত্ত কীমের উপর টাইটার স্থাপন করলে কীমের প্রতি
দিকের জন্য যে দর পাওয়া যায়, তাকে টাইটার দ্রবক বলে।

৬. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি।
এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও
কঠোরভাবে বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ
(0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে
কাজ করা হয়। ফলে পরিবেশ দূষণ উল্লেখযোগ্যভাবে কমে আসে।
অতএব, কোনো কোনো ক্ষেত্রে কম কষ্টের রাসায়নিক দ্রব্য ব্যবহৃত হয়।
ফলে পরিবেশ দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি
পরিবেশ বান্দব।

৭. উদাহরণের দ্রবণটি হলো সালফিউরিক এসিড (H_2SO_4) এর 10 M
দ্রব। H_2SO_4 একটি জারক ও ক্ষয়কারী পদার্থ। H_2SO_4 এর 10 M
দ্রব যখন পাত্রে H_2SO_4 দ্রবণ। পাত্রে H_2SO_4 পদার্থকে ক্ষয়কারী হ্যাঞ্জার
একভাবে অঙ্কন করা হয়। পাত্রে H_2SO_4 এর জন্য ক্ষয়কারী হ্যাঞ্জার
একটি নিচে অঙ্কন করে ব্যাখ্যা করা হলো:

প্রকৃতি	ক্ষয়কারী (Corrosive)
আয়তন	ক্ষয়কারী রাসায়নিক উপাদান কালে যে ক্ষয় রাসায়নিক দ্রব্যকে দ্রবের দর সম্পর্কে তুলে ও অন্য পদার্থের ক্ষয়কারী ক্ষতি সঞ্চিত হয়। এতে তুলে পুড়ে যায় এবং অন্যদিকে পদার্থ পুড়ে যায়।
উদাহরণ	পাত্রে কঠিন এসিডসলফিউরিক (HCl , HNO_3 , H_2SO_4) পাত্র দ্রব (NaOH , KOH , H_2O_2) আকসিজিন লিকার ক্রিডিং দ্রব ইত্যাদি।
সংরক্ষণ পদ্ধতি	<ul style="list-style-type: none"> পাত্র এসিডকে পোতলে রাখতে হবে। পাত্র ক্ষয়কারী, ক্ষয়কারী পদার্থের পিচ্ছিল পোতলে রাখতে হবে।

৮. উদাহরণের দ্রবণটি হলো 10M H_2SO_4 দ্রবণ এবং এর থেকে 1M
ঘনমাত্রার দ্রবণ প্রস্তুত করতে হবে 1ম ও ২য় চলকের ব্যবহার করতে
হবে।

1ম চলকের দ্রবণ হলো 100 ml আকসজনিতিক দ্রবণ ও এর দি হলো
নিম্নের। যেহেতু আকসজনিতিক দ্রবণের আয়তন 100 ml, তাই
আমাদের 100 ml 1M ঘনমাত্রার H_2SO_4 দ্রবণ তৈরি করতে হবে।
দ্রবণের সন্নিবেশ দ্রব হতে আনত।

$$V_1 S_1 = V_2 S_2$$

$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left(\frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 10ml H_2SO_4 দ্রবণের সঙ্গে পানি যোগ করতে হবে
(100-10) mL = 90 mL
আবার, নিম্নের দ্রবণে 10 ml দ্রব ঘনমাত্রার দ্রবণ নিচে তার
সঙ্গে আকসজনিতিক দ্রব 90 ml পানি যোগ করলে আমাদের
কামিত ঘনমাত্রার দ্রব পাওয়া যাবে।

চিত্র-১১



১. অক্সিজেনের দ্রবণ কী?
২. সাদা দ্রবক K, এর ঘন কখনো অসীম হয় না-ব্যাখ্যা
করে।
৩. A দ্রবের দ্রবীভূত উত্তর করার কৌশল বর্ণনা করে।
৪. B ও C দ্রবের সাথে সম্পর্কিত রাসায়নিক দ্রবণের
ব্যবহারকালে খান্দা কৃতি দ্রবণে কবীর উপর সম্পর্কে
বিশ্লেষণ করে। ৪
ও না প্রদেয় উত্তর

৫. অক্সিজেনের দ্রবণ একটি এক ধরনের রাসায়নিক বা কৃত্রিম দ্রবণ সঞ্চিত
বা দ্রব বা উপাদানসমূহকে জারণ প্রক্রিয়ায় জাল থেকে তুলে করে
খান্দা সঞ্চিত করে।

৬. সামান্য K, এর ঘন কখনো অসীম বা শূন্য হতে পারে না।
উদাহরণ ব্যাখ্যা করার জন্য নিম্নের উদাহরণী মাইক্রোপরি বিশ্লেষণ করি-
 $A + B \longrightarrow C + D$

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