

কুমিল্লা বোর্ড

১২। বিশুদ্ধ পানিতে কতক ঘেঁটা HCl মেল করলে K_w এর মান হবে কত?

- (ক) 1×10^{-16}
☒ (খ) 1×10^{-14}
 (গ) 1×10^{-13}
 (ঘ) 1×10^{-7}

১৩। কেন্দ্রীয় পরমাণুর অটক পূর্ণ হচ্ছে—

- i. H_2O
 ii. BCl_3
 iii. NCI_3

নিচের কোনটি সঠিক?

- (ক) i ও ii (খ) ii ও iii
☒ (গ) i ও iii (ঘ) i, ii ও iii

১৪। মানুষের রক্তের pH মান কত?

- ☒ (ক) 7-4
 (খ) 7-0
 (গ) 5-5
 (ঘ) 3-2

১৫। আয়নীকরণ বিভবের সঠিক ক্রম কোনটি?

- (ক) $N > O > F$
☒ (খ) $Br < Cl < F$
 (গ) $K < N < Li$
 (ঘ) $Be > Mg > C$

১৬। PH_4Cl বোঁলে বিদ্যমান বন্ধন—

- i. আয়নিক
 ii. সমযোজী
 iii. সল্লিবেশ

নিচের কোনটি সঠিক?

- (ক) i ও ii (খ) i ও iii
 (গ) ii ও iii ☒ (ঘ) i, ii ও iii

১৭। নিচের কোন বিক্রয়টি তিনপদার্থ?

- (ক) $HNO_3 + CH_3COOH$
☒ (খ) $CH_3COOH + H_2O$
 (গ) $HCl + CH_3COOH$
 (ঘ) $H_2SO_4 + CH_3COOH$

১৮। কোনটিতে পোল্যারাইজ বেনি ঘটেবে?

- ☒ (ক) $BeCl_2$
 (খ) $MgCl_2$
 (গ) $LiCl$
 (ঘ) $NaCl$

১৯। বরফের একটি অক্সিজেন পরমাণুতে কতটি H-বন্ধন বিদ্যমান?

- ☒ (ক) 2
 (খ) 3
 (গ) 4
 (ঘ) 5

২০। ^{28}Ni এর কতগুলো ইলেকট্রনের ক্ষেত্রে $(n + l) = 4$ হয় (এখানে, $l = 1$)?

- (ক) 4
 (খ) 5
☒ (গ) 6
 (ঘ) 7

২১। কোনটি দৃশ্যমান বর্ণালী?

- (ক) লাইম্যান সিরিজ
☒ (খ) বামার সিরিজ
 (গ) প্যাঞ্চেন সিরিজ
 (ঘ) ব্র্যাকট সিরিজ

২২। জাল পালসেট শনাক্তকরণে কোনটি ব্যবহার করা হয়?

- (ক) γ -রশ্মি (খ) X-রশ্মি
 (গ) IR-রশ্মি ☒ (ঘ) UV-রশ্মি

নিচের উদীপকটি পড় এবং ২৩ ও ২৪ নং প্রশ্নের উত্তর দাও :

বোঁল	ইলেকট্রন বিন্যাস	তড়িৎ ঋণাত্মকতা
L	ns^2	—
M	$(n+1)s^2$	—
N	ns^2sp^5	3.0

এখানে, $n = 3$

২৩। N_2 অণুর ক্ষেত্রে কোনটি প্রযোজ্য?

- (ক) দুক মোড় ইলেকট্রন একটি
☒ (খ) অপোলার সমযোজী
 (গ) পোলার সমযোজী
 (ঘ) sp সংকরণ বিশিষ্ট

২৪। LN_2 এবং MN_2 এর মধ্যে MN_2 অধিক—

- i. গলনাঙ্কবিশিষ্ট
 ii. পানিতে দ্রবণীয়
 iii. সমযোজী

নিচের কোনটি সঠিক?

- ☒ (ক) i ও ii (খ) ii ও iii
 (গ) i ও iii (ঘ) i, ii ও iii

২৫। সবচেয়ে দুর্বল এসিড কোনটি?

- ☒ (ক) H_3BO_3 (খ) HNO_3
 (গ) H_3PO_3 (ঘ) $HClO_3$

Ref: মাস্টারী স্মার
 Edition-2023; Page-637
 Ka এর মান $[5.8 \times 10^{-10}]$
 Ka এর মান $[1 \times 10^{-7}]$

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Mark Steyvers



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attended king william s college on the isle of man and then went to university college london to read english under albert
hugh smith his studies began in 1939 but were interrupted by the war in 1940 to be completed from 1945 to 1947

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