# Chemistry

#### FORMULAE SHEET

$n = \frac{m}{MM}$	$c = \frac{n}{V}$	PV = nRT
$q = mc\Delta T$	$\Delta G^{\circ} = \Delta H^{\circ} - T \Delta S^{\circ}$	$pH = -\log_{10}[H^*]$
$pK_{\alpha} = -\log_{10}[K_{\alpha}]$	$A = eIc = log_{10} \frac{I_o}{I}$	
Avogadro constant, N <sub>4</sub>		$6.022 \times 10^{23} \text{ mol}^{-1}$
Volume of 1 mole ideal g		
	at 0°C (273.15 K)	22.71 L
	at 25°C (298.15 K)	24.79 L
Gas constant		8.314 J mol <sup>-1</sup> K <sup>-1</sup>
Ionisation constant for water at 25°C (298.15 K), Kw		1.0 × 10 <sup>-14</sup>
Specific heat capacity of water		

#### DATA SHEET

#### Solubility constants at 25°C

Compound	$K_{\mu\nu}$	Compound	$K_{np}$
Barium carbonate	$2.58 \times 10^{-9}$	Lead(II) bromide	$6.60 \times 10^{-6}$
Barium hydroxide	$2.55 \times 10^{-4}$	Lead(II) chloride	$1.70 \times 10^{-5}$
Barium phosphate	$1.3 \times 10^{-29}$	Lead(II) iodide	$9.8 \times 10^{-9}$
Barium sulfate	$1.08 \times 10^{-10}$	Lead(II) carbonate	$7.40 \times 10^{-14}$
Calcium carbonate	$3.36 \times 10^{-9}$	Lead(II) hydroxide	$1.43 \times 10^{-15}$
Calcium hydroxide	$5.02 \times 10^{-6}$	Lead(II) phosphate	$8.0 \times 10^{-43}$
Calcium phosphate	$2.07 \times 10^{-29}$	Lead(II) sulfate	$2.53 \times 10^{-8}$
Calcium sulfate	$4.93 \times 10^{-5}$	Magnesium carbonate	$6.82 \times 10^{-6}$
Copper(II) carbonate	$1.4 \times 10^{-10}$	Magnesium hydroxide	$5.61 \times 10^{-12}$
Copper(II) hydroxide	$2.2 \times 10^{-20}$	Magnesium phosphate	$1.04 \times 10^{-24}$
Copper(II) phosphate	$1.40 \times 10^{-37}$	Silver bromide	$5.35 \times 10^{-13}$
Iron(II) carbonate	$3.13 \times 10^{-11}$	Silver chloride	$1.77 \times 10^{-10}$
Iron(II) hydroxide	$4.87 \times 10^{-17}$	Silver carbonate	$8.46 \times 10^{-12}$
Iron(III) hydroxide	$2.79 \times 10^{-39}$	Silver hydroxide	$2.0 \times 10^{-8}$
Iron(III) phosphate	$9.91 \times 10^{-16}$	Silver iodide	$8.52 \times 10^{-17}$
		Silver phosphate	$8.89 \times 10^{-17}$
		Silver sulfate	$1.20 \times 10^{-5}$

# **Hsc Chemistry Formula Sheet**

**JE Gale** 

**Hsc Chemistry Formula Sheet:** 

Unveiling the Magic of Words: A Review of "Hsc Chemistry Formula Sheet"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Hsc Chemistry Formula Sheet**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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