

MODEL SOLUTIONS TO IIT JEE 2010


Paper I - (Code 0)

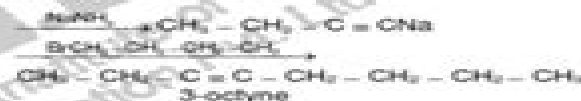
PART I

		1 D	2 A	3 B	4 B	5 C	6 C	7 D	8 A		
	9 B, D		10 B, C		11 A, B		12 C, D		13 A, C		
			14 B	15 D	16 C	17 B	18 C				
19 3	20 2	21 0	22 3	23 4	24 3	25 3	26 1	27 0	28 5		

Section I




- k varies exponentially with T.
- Standard enthalpy of formation of elements in their standard state is taken as zero.
- Ligand NO_2^- and counter ion Cl^- are mutually exchanged.
- EDTA is 
- Bond energy of C-C is 83 Kcal mol^{-1}
- $\text{CH}_3 - \text{CH}_2 - \text{C} \equiv \text{CH}$



8. Ring (a) is pyranose with α -glycosidic link.

Section II

- 
2, 2-dimethyl butane
- $\text{Ca}(\text{OH})_2$ and Na_2CO_3 can be used for softening temporary hard water.
- Molar conductivity and electromotive force are intensive properties.
- HNO_3 and $\text{CH}_3\text{COONa} \rightarrow \text{HNO}_2$ on mixing with excess CH_3COONa also results in a mixture of CH_3COOH and CH_3COONa .

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