Chemistry 338 2nd Semester Final Exam Review Sheet

Unit 1 - Matter & Measurement - Chapters 1 & 2

 A student experimentally determines the density of an object to be 4.2 g/mL. If the accepted value is 5.0 g/mL, calculate the % error.

% error = $\frac{experimentat-accepted}{accepted} \times 100$ % error = $\frac{4.2-5.0}{5.0} \times 100 = -16\%$ error

Unit 2 - Periodic Table - Chapter 5

2.

Element	Symbol	Atomic #	Protons	Neutrons	Electrons	Mass#	Charge
Fungsten ion	194 W+2	7-4	74	110	72	184	+2

 Using the following atomic radius values, predict the atomic radius of potassium (K): Li = 145 pm, Na = 180 pm, K = 7, Rb = 235 pm K = 220 pm

Unit 3 - Reactions/Compounds - Chapters 7 & 8

- 4. Classify the compound as ionic or covalent. Name the compound
 - Fe₂O₃ iron (III) oxide (ionic roman numerals for transition metals, change ending to -ide)
 - b. Cl₂O₈ dichlorine octaoxide (covalent use prefixes, change ending to -ide)
- 5. Classify the compound as ionic or covalent. Write the formula
 - a. Trisulfur Dinitride S₃N₂ (covalent)
 - b. Zinc Hydroxide Zn(OH)2 (ionic look up ion charges and criss-cross) Zn⁻² OH⁻³
 - c. Chromium (III) nitrate Cr(NO₃)₃ (ionic criss-cross) Cr²³ NO₃⁻¹
- 6. Write the equation, balance it, and classify the type of reaction:

Iron (II) sulfate reacts with sodium hydroxide to yield iron(II) hydroxide and sodium sulfate

> 1 FeSO₄ + 2 NaOH → 1 Fe(OH)₂ + 1 Na₂SO₄ double displacement (two ionic compounds reacting)

7. Balance the equation and classify the type of reaction:

1 Cu + 2 AgNO₃ → 1 Cu(NO₃)₂ + 2 Ag Single displacement (element and a compound reacting)

Unit 4 - Moles - Chapter 3

8. Calculate the molar mass of Fe₂(SO₄)₅.

Fe (2)(55.9 g/mol) = 111.8 g/mol

S (3)(32.1 g/mol) = 96.3 g/mol

+ O (12)(16.0 g/mol) = +192 g/mol

= 400.1 g/mol

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beliefs of our readers across the political spectrum Blended Learning: Concepts, Methodologies, Tools, and Applications Management Association, Information Resources, 2016-08-18 Traditional classroom learning environments are quickly becoming a thing of the past as research continues to support the integration of learning outside of a structured school environment Blended learning in particular offers the best of both worlds combining classroom learning with mobile and web based learning environments Blended Learning Concepts Methodologies Tools and Applications explores emerging trends case studies and digital tools for hybrid learning in modern educational settings Focusing on the latest technological innovations as well as effective pedagogical practice this critical multi volume set is a comprehensive resource for instructional designers educators administrators and graduate level students in the field of education Correspondence Courses Offered by Colleges and Universities Through the United States Armed Forces Institute, 1957 Research in Education ,1974 <u>Undergraduate Announcement University of Michigan--Dearborn, 1983</u> **Announcement** University of Michigan--Dearborn, 1981 Chemical and Engineering News ,1959 **DOD Pam** United States. Office of Armed Forces Information and Education, 1957 Official Gazette Philippines, 1999 Department of the Army Pamphlet, 1951 Annual Report of Illinois State Board of Health Illinois State Board of Health, 1894

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