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the positions of the peroxy radical and the transferred h atom and the types of carbons from which the h atom

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same index and radicand are known as like radicals it is often helpful to treat radicals just as you would treat variables like
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requires significant amounts of energy for example splitting H_2 into 2H has a Δh of 435 kJ/mol and Cl_2 into 2Cl has a Δh of

243 kJ/mol this is known as the homolytic bond dissociation energy and is usually abbreviated as the

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web oct 6 2021 answer 5 3 2 notice that the terms involving the square root in the denominator are eliminated by multiplying by the conjugate we can use the property $a/b \cdot a/b = a^2/b^2$ to expedite the process of multiplying the expressions in the denominator example 5 4 16 rationalize the denominator 10 2 6

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