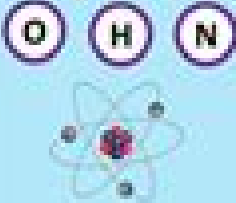
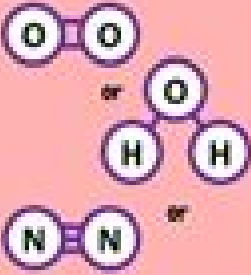
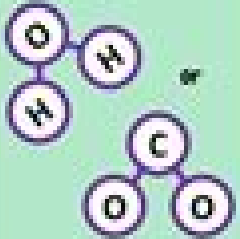


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Atoms – Elements – Molecules – Compounds

Atoms	Elements	Molecules	Compounds
The smallest unit of matter and has three main parts (protons, neutrons, and electrons). The basic building blocks of matter. Smallest part of an element.	A pure substance consisting of one type of atom and can be identified by its atomic number (number of protons). All elements are found on the Periodic Table of Elements and have a chemical symbol.	Formed when two or more atoms join together chemically. Can be the same atoms.	Formed when two different elements join together. All compounds are molecules, but all molecules are not compounds.
			
Example Oxygen, Hydrogen, Nitrogen	Example Oxygen's atomic number is 8	Example Oxygen, Water, Nitrogen, Carbon Dioxide	Example Water, Carbon Dioxide

Atoms, Elements, Molecules and Compounds

Atoms Molecules Elements Matter Energy

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