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FOCUS 2

MOLECULES

2A.2 (a) 5; (b) 8; (c) 12; (d) 7

2A.4 (a) $[\text{Ar}] 3d^1$; (b) $[\text{Ar}] 3d^5$; (c) $[\text{Kr}] 4d^{10}$; (d) $[\text{Xe}]$

2A.6 (a) $[\text{Kr}]$; (b) $[\text{Xe}] 4f^{14}5d^5$; (c) $[\text{Xe}]$; (d) $[\text{Ar}]$

2A.8 (a) Ca: $[\text{Ar}] 4s^2$; Ti^{2+} : $[\text{Ar}] 3d^2$; V^{3+} : $[\text{Ar}] 3d^2$ In the d block, the energies of the $n-1$ d-orbitals lie below those of the ns-orbitals. Therefore, when V and Ti form ions they lose their 3s electrons before losing their 4d electrons. (b) Ca: no unpaired electrons; Ti^{2+} : two unpaired electrons; V^{3+} : two unpaired electrons. (c) Ti^{2+} : $[\text{Ar}] 3d^1$ no neutral atom has this electron configuration.

2A.10 (a) Au^{3+} ; (b) Os^{3+} ; (c) I^{3+} ; (d) As^{3+}

2A.12 (a) Cr^{2+} ; (b) Ag^{2+} ; (c) Zn^{2+} ; (d) Pb^{2+}

2A.14 (a) 4d; (b) 3p; (c) 5p; (d) 4s

2A.16 (a) +1; (b) -2; (c) +2; (d) -3; (e) -1

Atkins 7th Edition Solutions Manual

Carl Hoeger, Laurence Lavelle, Yinfu Ma



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US Solutions Manual to Accompany Elements of Physical Chemistry 7e David (Faculty Education Director and Undergraduate Dean for the Faculty of Science Smith, and Deputy Head of the School of Chemistry Bristol University), David Smith, 2017-10-05 The Solutions Manual to Accompany Elements of Physical Chemistry 7th edition contains full worked solutions to all end of chapter discussion questions and exercises featured in the book

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these questions , *Student Solutions Manual for Physical Chemistry, Seventh Edition* Peter Atkins,2002

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