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Chapter 2

Atomic Structure

2-6

- (a) Aluminum foil used for storing food weighs about 0.3 g per square cm. How many atoms of aluminum are contained in one 6.25 cm² size of foil?
- (b) Using the densities and atomic weights given in Appendix A, calculate and compare the number of atoms per cubic centimeter in (i) lead and (ii) lithium.

Solution: (a) In a 6.25 cm² sample:

$$\frac{(0.3 \text{ g}) \left(6.022 \times 10^{23} \text{ atoms/mol} \right)}{(26.981 \text{ g/mol})} = 6.7 \times 10^{21} \text{ atoms}$$

(b) (i) In lead:

$$\frac{(11.36 \text{ g/cm}^3) \left(6.022 \times 10^{23} \text{ atoms/mol} \right)}{(207.19 \text{ g/mol})} = 3.30 \times 10^{22} \text{ atoms/cm}^3$$

(ii) In lithium:

$$\frac{(0.534 \text{ g/cm}^3) \left(6.022 \times 10^{23} \text{ atoms/mol} \right)}{(6.94 \text{ g/mol})} = 4.63 \times 10^{22} \text{ atoms/cm}^3$$

Despite the different mass densities of Pb and Li, their atomic densities are approximately the same.

2-7

- (a) Using data in Appendix A, calculate the number of iron atoms in one ton (1000 kg) of iron.
- (b) Using data in Appendix A, calculate the volume in cubic centimeters occupied by one mole of boron.

Solution: (a)

$$\frac{(1000 \times 10^3) \left(6.022 \times 10^{23} \text{ atoms/mol} \right)}{(55.847 \text{ g/mol})} = 1.078 \times 10^{28} \text{ atoms}$$

Askeland Solution Manual Polymers

Guillaume Favre



Askeland Solution Manual Polymers:

This book delves into Askeland Solution Manual Polymers. Askeland Solution Manual Polymers is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Askeland Solution Manual Polymers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Askeland Solution Manual Polymers
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 - Chapter 3: Askeland Solution Manual Polymers in Everyday Life
 - Chapter 4: Askeland Solution Manual Polymers in Specific Contexts
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