

- d. Nope, you will have to ask our petty cash custodian for your reimbursement. All disbursements below \$5,000 will have to be paid out of the petty cash fund.
8. By requiring that all deposits must be deposited intact and all disbursements, except petty disbursements, be made through check, you are implementing an internal control called
- Regular bank reconciliations.
 - Imprest system.
 - Voucher system.
 - Bossy system.
9. By requiring that all disbursements must be supported by CDVs, you are implementing an internal control called
- Regular bank reconciliations.
 - Imprest system.
 - Voucher system.
 - Bossy system.
10. Which of the following is incorrect regarding the petty cash fund?
- The petty cash fund should only be used for recurring expenses that are not material enough to warrant disbursement through check.
 - Each petty cash disbursement must be supported by a properly authorized petty cash voucher.
 - The requesting party should affix his or her signature on the petty cash voucher acknowledging the receipt of the petty cash.
 - For a big company, each petty cash voucher must be authorized by the Chief Executive Officer or President.

Chapter 9

Accounting Practice Set

Learning Competencies

The learners should be able to.....

1. Perform the steps in the accounting cycle, from preparation of documents to the preparation, analysis, and interpretation of financial statements

Practice Set

The November 30, 20x1 trial balance of Rak N' Koll Co., a call center business, shows the following information:

RAK N' KOLL

Trial Balance
November 30, 20x1

ACCOUNTS	Debit	Credit
Cash	340,000	
Accounts receivable		
Prepaid insurance		
Computer equipment	1,400,000	
Accumulated depreciation		240,000
Loans payable		1,000,000
Owner's equity		1,254,000
Owner's drawings	3,300,000	
Service fees		4,890,000
Salaries expense	990,000	
Rent expense	1,100,000	
Utilities expense	165,000	
Office supplies expense	84,000	
Insurance expense	5,000	
Totals	7,384,000	7,384,000

Accounting Practice Set Solutions

David Hawkrige, Roland Kaye



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Intermediate Accounting Terry D. Warfield, Jerry J. Weygandt, Donald E. Kieso, 2007-12-04 Now readers can get all the accuracy and authority of the best selling intermediate accounting book in the new second edition of this brief streamlined version Fundamentals of Intermediate Accounting presents a balanced discussion of concepts and applications explaining the rationale behind business transactions before addressing the accounting and reporting for those activities Readers will gain a solid foundation in such areas as the standard setting process the three major financial statements revenue recognition income taxes reporting disclosure issues and much more

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The Accounting Educators' Journal, 1988

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Business Education Forum ,1984 *DOD Pam* United States. Office of Armed Forces Information and Education,
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web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1 10 the capacity of the syringe the 20 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the syringe to the connector at the end of sensor

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the

standard molar volume figure 9 18

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web this lab explores the relationships between pressure volume and temperature in an ideal gas the ideal gas law $pV = nRT$ is used and with it we can explore it by seeing how the pressure and volumes changed in the tubes each time that we

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web figure pageindex 10 since the number of moles in a given volume of gas varies with pressure and temperature changes chemists use standard temperature and pressure 273 15 K and 1 atm or 101 325 kPa to report properties of gases

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