

BALANCING CHEMICAL EQUATIONS

Load the simulation *Balancing Chemical Equations*

<http://phet.colorado.edu/en/simulation/balancing-chemical-equations>

CRITICAL THINKING QUESTIONS

- Explore the *Balancing Chemical Equations* simulation. Discuss with your group what you find.
 - What are the different ways that the simulation indicates when an equation is balanced (look in the bottom left corner of the simulation to see the options)?

- For each **balanced** reaction, indicate the total number of molecules in the table below. You will have to try to balance each of the equations.

Reaction	Total Number of Molecules	
	Reactant Side (Left)	Product Side (Right)
Make Ammonia		
Separate Water		
Combust Methane		

- Is the number of total molecules on the left side of a balanced equation always equal to the number of total molecules on the right side of the equation? Explain your answer.

- For each balanced reaction, indicate the total number of atoms in the table below.

Reaction	Total Number of Atoms	
	Reactant Side (Left)	Product Side (Right)
Make Ammonia		
Separate Water		
Combust Methane		

- Is the number of total atoms on the left side of a balanced equation always equal to the number of total atoms on the right side of the equation?
- What is the same on the left and right side of a balanced equation? Explain your answer.
- As a group, play level 1 of the balancing equation game. Write down the strategies your group uses to balance chemical equations.

Balancing Chemical Equations Phet Lab Tablet Compatible Answers

J Dewey



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