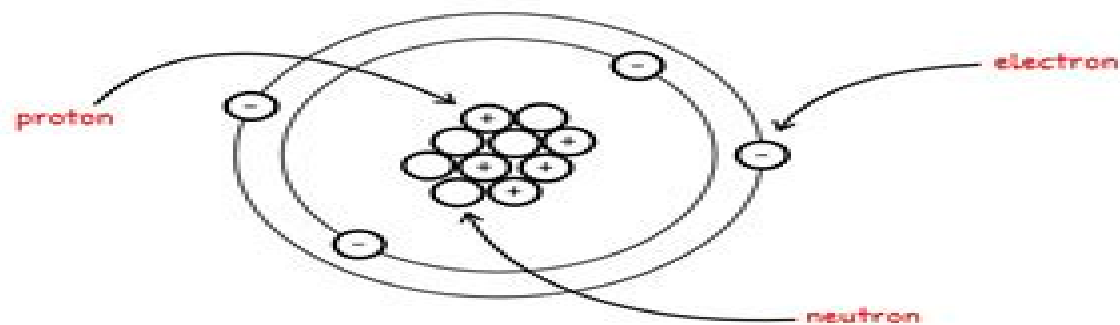


Name answer key

Date _____

Atomic Structure

Label the parts of an atom on the diagram below, and answer the questions!



1. What type of charge does a proton have?
Positive (+1)
2. What type of charge does a neutron have?
Neutral (0)
3. What type of charge does an electron have?
Negative (-1)
4. Which two subatomic particles are located in the nucleus of an atom?
Protons and neutrons
5. If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus?
35
6. What is the atomic number of the atom in the diagram above?
5
7. What is the atomic mass/mass number of the atom in the diagram above?
(11+11=22) *protons and neutrons added together) 22 is the atomic mass/mass number
8. How many protons are in the nucleus of an atom with an atomic number of 15?
15 *number of protons is the same as the atomic number
9. How many electrons are in the nucleus of an atom with an atomic number of 20?
20
10. How many neutrons are in the nucleus of an atom with an atomic number of 25?
(use Periodic Table for mass) 30 Neutrons. atomic mass: 54.938 ~ 55 atomic number: 25.....55-25=30
11. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?
Mass number = 6.9% (Li-Lithium has an atomic number of 3 which means it has 3 protons.
12. How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?
Neutrons = 11 (atomic mass: 36 - 25:atomic number)

Atomic Structure Guided Practice Problems Answer Key

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1. Understanding the eBook Atomic Structure Guided Practice Problems Answer Key
 - The Rise of Digital Reading Atomic Structure Guided Practice Problems Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Guided Practice Problems Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure Guided Practice Problems Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Guided Practice Problems Answer Key
 - Personalized Recommendations
 - Atomic Structure Guided Practice Problems Answer Key User Reviews and Ratings
 - Atomic Structure Guided Practice Problems Answer Key and Bestseller Lists
5. Accessing Atomic Structure Guided Practice Problems Answer Key Free and Paid eBooks
 - Atomic Structure Guided Practice Problems Answer Key Public Domain eBooks
 - Atomic Structure Guided Practice Problems Answer Key eBook Subscription Services
 - Atomic Structure Guided Practice Problems Answer Key Budget-Friendly Options

6. Navigating Atomic Structure Guided Practice Problems Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Atomic Structure Guided Practice Problems Answer Key Compatibility with Devices
 - Atomic Structure Guided Practice Problems Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Atomic Structure Guided Practice Problems Answer Key
 - Highlighting and Note-Taking Atomic Structure Guided Practice Problems Answer Key
 - Interactive Elements Atomic Structure Guided Practice Problems Answer Key
8. Staying Engaged with Atomic Structure Guided Practice Problems Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Structure Guided Practice Problems Answer Key
9. Balancing eBooks and Physical Books Atomic Structure Guided Practice Problems Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Structure Guided Practice Problems Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Atomic Structure Guided Practice Problems Answer Key
 - Setting Reading Goals Atomic Structure Guided Practice Problems Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Atomic Structure Guided Practice Problems Answer Key
 - Fact-Checking eBook Content of Atomic Structure Guided Practice Problems Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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