NEET 2020 Paper



- 59. A tertiary butyl carbocation is more stable than a secondary butyl carbocation because of which of the following?
 - (1) -R effect of -CH, groups
 - (3) -I effect of -CH, groups

9 a - H Hyperconjugation

- (2) Hyperconjugation
- (4) + R effect of -CH, groups

- CH,—C—CH,
 CH,
 Tert butyl carbocation
 stable due to
- CH,—CH—CH,—CH,
 sec. butyl carbocation
 Stable due to
 5 α H Hyperconjugation
- 60. Find out the solubility of Ni(OH), in 0.1 M NaOH. Given that the ionic product of Ni(OH), is 2×10⁻¹⁵.
 - (1) $1 \times 10^{-13} \text{ M}$

(2) 1 × 10° M (4) 2 × 10° M

(3) $2 \times 10^{-13} \text{ M}$

Sol.

Sect.

- 3 Ni(OH), ← Ni²⁺
 - Ni²⁺ + 2OH² s 0.1 M

∴
$$K_{so} = [Ni^{2}] [OH^{-}]^{2}$$

⇒ $k_{so} = s \times (0.1)^{2}$
 $\frac{2 \times 10^{-15}}{3}$

$$\Rightarrow s = \frac{2 \times 10^{-13}}{0.1^3}$$

$$\Rightarrow s = 2 \times 10^{-13} \text{ M}$$

- 61. Reaction between acetone and methylmagnesium chloride followed by hydrolysis will give:
 - (1) Tert, butyl alcohol

(2) Isobutyl alcohol

(3) Isopropyl alcohol
Sol. 1

(4) Sec. butyl alcohol

Tertbutylalcohol

O OM

62. Which of the following amine will give the carbylamine test?

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Xiang Xie

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