

Advanced Theory practice Test

Multiple choice (single answer) to the three main components the question:

1. (1.1.1.1) What is the correct order of the three components?
- By identifying the number of protons
 - By identifying the number of electrons
 - By calculating the number of protons/neutrons
 - By identifying the number of isotopes

2. (1.1.1.2) What is the correct order of the three components?
- By identifying the number of protons
 - By identifying the number of electrons
 - By calculating the number of protons/neutrons
 - By identifying the number of isotopes

3. (1.1.1.3) Which of the following is the correct order of the three components?
- By identifying the number of protons
 - By identifying the number of electrons
 - By calculating the number of protons/neutrons
 - By identifying the number of isotopes



4. (1.1.1.4) What is the correct order of the three components of the three components of the three components?
- By identifying the number of protons
 - By identifying the number of electrons
 - By calculating the number of protons/neutrons
 - By identifying the number of isotopes

5. (1.1.1.5) Which of the following is the correct order of the three components of the three components of the three components?

Figure 1:



6. (1.1.1.6) Which of the following is the correct order of the three components of the three components of the three components?
- By identifying the number of protons
 - By identifying the number of electrons
 - By calculating the number of protons/neutrons
 - By identifying the number of isotopes

7. (1.1.1.7) Which of the following is the correct order of the three components of the three components of the three components?

- By identifying the number of protons
- By identifying the number of electrons
- By calculating the number of protons/neutrons
- By identifying the number of isotopes

Atomic Theory Practice Test

Arshad Iqbal

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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