Multiple Choice: (from Zumdahl Chemistry, 9th Edition, AP Edition)

Refer to the following diagram for questions 1-3

A 100.0 mL Graduated Cylinder is filled with 100.0 mL of water. A few drops of food colouring are added to the top of the water, without mixing. Initially, the top of the graduated cylinder appears to be very dark, while the bottom remains clear and colourless. After 5 min, the liquid in the graduated cylinder appears to be uniformly coloured.

- Classify the contents in the graduated cylinder before any food colouring is added.
- (A) Element (B) Compound. (C) Heterogeneous mixture
- (D) Homogeneous mixture
- Classify the contents in the graduated cylinder just after the food colouring is added.
- (A) Element (B) Compound. (C) Heterogeneous Mixture (D) Homogeneous mixture
- 3. Classify the contents in the graduated cylinder after the 5 minutes has elapsed.
- (A) Element (B) Compound. (C) Heterogeneous Mixture (D) Homogeneous mixture
- 4. A 10.00 g piece of metal is submerged in a graduated cylinder initially containing 20.00 mL of water. With the metal in the cylinder, the level of water is recorded as 21.40 mL. Determine the identity of the metal.
- (A) Al ($\rho = 2.70 \text{ g/mL}$)
- (B) $Zn (\rho = 7.13 \text{ g/mL})$
- (C) Ag ($\rho = 10.49$ g/mL)
- (D) Au ($\rho = 19.32 \text{ g/mL}$)
- 239 naPu → X + a

In the decay represented above, what is the nucleon number of particle X?

- (A) 92 (B) 96 (C) 235 (D) 243
- 6. Which of the following statements about atoms is CORRECT?
 - (A) Atoms are charged because they have different numbers of protons and neutrons>
 - (B) All atoms of an element have the same number of nucleons and electrons.
 - (C) An atom is mostly filled with protons.
 - (D) The nucleus is negatively charged.
- When each of the following is heated to 50°C and the temperature is held constant for 5 min, which one undergoes only a physical change.
 - (A) egg (B) steak (C) ice cream (D) cake batter
- 7. A sample of Ti was ionized by electron impact in a mass spectrometer. Information from the mass spectrum about the isotopes of Ti in the sample is shown in Table 2.

Table 1. Results from mass spectrum of Ti sample				
me/z	46	47	48	49
Percent Abundance	9.1	7.8	74.6	8.5

BIEFORE.

(DATAMENTO)

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APTER

Calculate the relative atomic mass of Ti in this sample. Report answers to one decimal place.

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N Noddings



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web include the earthing resistance and the minimum size of the earthing conductor which can be calculated using various methods as part of the design methodology lim s c and al shawesh y 35 proposed a systematic step by step approach for the

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web various types of earthing are mentioned below plate earthing pipe earthing strip earthing plate earthing the most efficient way of earthing is plate earthing which involves installing a galvanized copper plate 3m deep in

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web plate earthing fig 3 plate earthing for this form of earthing a plate composed of copper or galvanized iron is placed vertically in the ground pit less than three meters above the ground for a more effective electrical grounding system one must maintain the earth's moisture condition surrounding the plate earthing system

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