Electron Configurations Worksheet - Answers

Write the complete ground state electron configurations for the following:

1)	lithium	1s ² 2s ¹
2)	oxygen	1s ² 2s ² 2p ⁴
3)	calcium	1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 4s ²
4)	titanium	1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 4s ² 3d ²
5)	rubidium	1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 4s ² 3d ¹⁰ 4p ⁶ 5s ¹
6)	lead	1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 4s ² 3d ¹⁰ 4p ⁶ 5s ² 4d ¹⁰ 5p ⁶ 6s ² 4f ¹⁴ 5d ¹⁰ 6p ²
7)	erbium	1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 4s ² 3d ¹⁰ 4p ⁶ 5s ² 4d ¹⁰ 5p ⁶ 6s ² 4f ¹²

Write the abbreviated ground state electron configurations for the following:

8)	helium	1s ² (this one cannot be abbreviated)	
9)	nitrogen	[He] 2s ² 2p ³	
10)	chlorine	[Ne] 3s ² 3p ⁵	
11)	iron	[Ar] 4s ² 3d ⁶	
12)	zinc	[Ar] 4s ² 3d ¹⁰	
13)	barium	[Xe] 6s ²	
14)	polonium	[Xe] 6s ² 4f ¹⁴ 5d ¹⁰ 6p ⁴	

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T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder

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