

Study Buddy: Atomic Structure and Nuclear Chemistry

Development of the Modern Atomic Theory

Scientist	Experiment	Findings	Limitations	Model
Democritus	No experiment	Atoms are the smallest form of matter and cannot be created, destroyed or divided	Did not explain chemical behavior	
Dalton	Use experiments of previous scientists	Postulates: Atoms of elements are identical; atoms can't be created or destroyed; atoms combine in whole number ratios in a chemical reaction	Did not include sub-atomic particles	
Thomson	Cathode ray	Discovered the electron "Plum-pudding" Model	Did not explain deflection pattern of alpha particles as seen by Rutherford	
Rutherford	Gold foil	Discovered the proton and nucleus. Atoms contain mostly empty space.	Could not explain why electrons do not crash into the nucleus	
Chadwick	Alpha particles	Discovered the neutron	No modification of atom model	No new model/neutron later add to nucleus
Bohr	Uses findings from previous scientists and Quantum theory	Electrons are in orbits of differing energy levels.	Does not explain the behavior of other elements just hydrogen	
Modern Atom	Multiple experiments	Electron cloud and dual nature of electrons	Still doesn't explain the exceptions.	

Parts of an Atom (8th Grade Review)

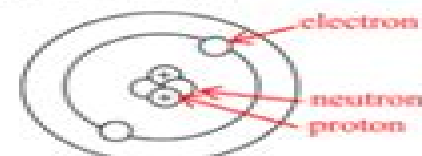
Sub-atomic Particle	Location	Charge	Mass
Proton (p ⁺)	Nucleus	1+	1
Neutron (n ⁰)	Nucleus	0	1
Electron (e ⁻)	outside the nucleus	1-	0 or 1/1840

Nuclear Symbol Carbon-14



Protons: 6
 Neutrons: 8
 Atomic Number: 6
 Mass Number: 14

Identify the proton, electron, neutron in the model below.



What is the atomic number of the above models? 2
 What is the atomic mass? 4.00 amu
 What element is this? Helium

Atomic Structure Study Guide And Nuclear Processes

**United States. Congress. Senate.
Committee on Foreign
Relations, United States**



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Study Material of Electrical Power Systems for GATE & ESE (Theory & Questions) Volume 1 Umesh

Dhande,2024-11-07 This comprehensive guide is designed to cater to the growing demand for accurate and concise concepts and formulas for electrical engineering power systems Vol 1 The book s key features include 1 Step by Step Solutions Detailed easy to follow solutions to all questions 2 Chapter Wise and Year Wise Analysis In depth analysis of questions organized by chapter and year 3 Detailed Explanations Clear explanations of each question ensuring a thorough understanding of the concepts 4 Simple and Easy to Understand Language Solutions are presented in a straightforward and accessible manner

General Chemistry Workbook Daniel C. Tofan,2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material

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Environment Peter H. Raven,Linda R. Berg,David M. Hassenzahl,2012-12-17 Raven s 8th edition of Environment offers more detailed content than the Visualizing text for a better understanding and integration of the core environmental systems and to view and analyze the role those systems play Shorter but still comprehensive coverage focuses on ethical decision making and key local environmental science issues requiring readers to think critically about the course material outside of the classroom Other features include brief text in the comprehensive segment extensive chapter pedagogy to help reinforce the systems approach more opportunities to think critically about the how systems intersect and fit together and new data interpretation questions at the end of each chapter

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