

6.14. Density:

6.14.1. Definition:

1.000 liters of pure water (H₂O) corresponds to 1.000 kg. That is a nice convenience, but does not apply to all other liquids, different in composition from pure water (H₂O): for some liquids, a 1.000 kg of that liquid will weigh more, and for some liquids, a 1.000 kg will weigh less. Obviously, the type of liquid will have an influence, this is how got defined the specific weight, or density of a type of matter/liquid/gas:

$$\text{DENSITY} = \frac{\text{MASS}}{\text{VOLUME}} \quad \text{OR} \quad \rho = m/v$$

Density describes the ratio between the mass of a body of matter, or a liquid on the one hand and the volume it takes up, on the other hand. The symbols and unit of density:

$$\text{DENSITY } [\rho] = \frac{\text{MASS}}{\text{VOLUME}} \quad \frac{[m]}{[V]}$$

$$\text{DENSITY } [\rho] = \frac{\text{MASS}}{\text{VOLUME}} \quad \frac{[kg]}{[m^3]}$$

Density is expressed by the Greek symbol "ρ", pronounced as "rho". The unit in the metric system is expressed in KILOGRAM/M³ and helps to express the mass of 1 m³, of a certain substance (solid, liquid, or gaseous). Density is one of the objective criteria to distinguish substances from each other. For example:

EXAMPLE 6.14.1: AvGas (Aviation Gasoline) has a density of 0,72 g/cm³.

How many kg/m³ does that correspond with?

- a. 72.000 kg/m³
- b. 720 kg/m³
- c. 72 kg/m³
- d. 7.200 kg/m³



ANSWER:

Here we are facing a conversion of 2 factors, both the mass and the volume, before we can state anything useful with regards to what was asked.

STEP 1: Let's first tackle the mass, which needs to be converted to the standard unit: g -> kg:

kilogram	hectogram <i>(rarely used unit)</i>	decagram <i>(rarely used unit)</i>	gram	decigram
0,	0	0	7	2

72 g corresponds to 0,0072 kg

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O García



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Airline Transport Pilot and Aircraft Dispatcher Written Test Book ,1991 **Flight and Ground Instructor Written Test Book** ,1991 **Air Force Manual** United States. Department of the Air Force,1976 **Flight Engineer Written Test Book** ,1991 *Principles of Flight for Pilots* Peter J. Swatton,2011-06-24 Organised and written as an accessible study guide for student pilots wishing to take commercial ground examinations to obtain ATPL or CPL licenses Principles of Flight for Pilots also provides a reliable up to date reference for qualified and experienced personnel wishing to further improve their understanding of the Principles of Flight and related subjects Providing a unique aerodynamics reference tool unlike any book previously Principles of Flight for Pilots explains in significant depth all the topics necessary to pass the Principles of Flight examination as required by the EASA syllabus Aviation ground instructor Peter J Swatton well reputed for his previous works in the field of pilot ground training presents the subject in seven parts including basic aerodynamics level flight aerodynamics stability manoeuvre aerodynamics and other aerodynamic considerations Each chapter includes self assessed questions 848 in total spread over eighteen chapters with solutions provided at the end of the book containing full calculations and explanations **Profile** ,1990 **Profile. United States Department of Defense** ,1981 **Resources in Education** ,1990 **Flying Magazine** ,1972-03 Instrument Rating Written Test Book ,1992

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