#### **ACTIVITY SERIES WORKSHEET ANSWERS**

#### Complete the table below using the activity series of metals

	Na	Zn	Al	Cu
	Na + CuCl <sub>2</sub>	Zn + CuCl <sub>2</sub>	Al + CuCl <sub>2</sub>	
	<b>→</b>	<b>→</b>	<b>→</b>	NR
CuCl <sub>2</sub>	NaCl <sub>2</sub> + Cu	ZnCl <sub>2</sub> + Cu	AlCl <sub>3</sub> + Cu	
	Na + Fe(NO <sub>3</sub> ) <sub>3</sub>	Zn + Fe(NO <sub>3</sub> ) <sub>3</sub>	Al + Fe(NO <sub>3</sub> ) <sub>3</sub>	
	<b>→</b>	<b>→</b>	<b>→</b>	NR
Fe(NO <sub>3</sub> ) <sub>3</sub>	NaNO <sub>3</sub> + Fe	Zn(NO <sub>3</sub> ) <sub>2</sub> + Fe	Al(NO <sub>3</sub> ) <sub>3</sub> + Fe	
	Na + AgNO <sub>3</sub>	Zn + AgNO <sub>3</sub>	Al + AgNO <sub>3</sub>	
	→	→	<b>→</b>	NR.
	NaNO <sub>3</sub> + Ag	Zn(NO <sub>3</sub> ) <sub>2</sub> + Ag	Al(NO <sub>3</sub> ) <sub>3</sub> + Ag	
	Na + H <sub>2</sub> SO <sub>4</sub>	Zn + H <sub>2</sub> SO <sub>4</sub>	AI + H <sub>2</sub> SO <sub>4</sub>	
	$\rightarrow$	→	→	NR
H₂SO <sub>4</sub>	Na <sub>2</sub> SO <sub>4</sub> + H <sub>2</sub>	ZnSO <sub>4</sub> + H <sub>2</sub>	AJ <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub> + H <sub>2</sub>	
	Na + Al(OH) <sub>1</sub>			
	→	NR	NR	NR
Al(OH) <sub>3</sub>	NaOH + Al			

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Applying The Activity Series Worksheet Answers**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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