

ATOMIC STRUCTURE QUIZ



Multiple Choice: Read the question carefully and select the choice that best answers the question. 2 pt each

1.) The charge of a proton is...

- a.) Negative
- b.) Positive
- c.) Neutral
- d.) Unknown

2.) What does the atomic number tell you?

- a.) Number of neutrons
- b.) The size of the atom
- c.) The mass of the atom
- d.) The number of protons

3.) How can you find the mass of an isotope?

- a.) Adding protons and neutrons
- b.) Adding protons and electrons
- c.) Subtracting protons and neutrons
- d.) Subtracting electrons and neutrons

4.) What is the Atomic Mass of Lithium?

- a.) 6.94
- b.) 3
- c.) 4
- d.) 6.023

5.) The charge of an electron is...

- a.) Negative
- b.) Positive
- c.) Neutral
- d.) Unknown

6.) In a Bohr model, how many electrons can go in the first orbital?

- a.) 8
- b.) 16
- c.) 2
- d.) 3

7.) What is the element symbol for Radon?

- a.) Rb
- b.) Ra
- c.) Rn
- d.) Ru

8.) Which subatomic particle does not have a mass?

- a.) Neutr
- b.) Proton
- c.) Neutron
- d.) Electron

9.) In a neutral atom, the number electrons equals the number of...

- a.) Neutrons
- b.) Protons
- c.) Mass
- d.) Cannot Determine

10.) What is the name of the element with the symbol, Hg

- a.) Mercury
- b.) Bromine
- c.) Iodine
- d.) Helium

Atomic Structure Test Questions

RJ Alexander

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Table of Contents Atomic Structure Test Questions

1. Understanding the eBook Atomic Structure Test Questions
 - The Rise of Digital Reading Atomic Structure Test Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Test Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure Test Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Test Questions
 - Personalized Recommendations
 - Atomic Structure Test Questions User Reviews and Ratings
 - Atomic Structure Test Questions and Bestseller Lists
5. Accessing Atomic Structure Test Questions Free and Paid eBooks

- Atomic Structure Test Questions Public Domain eBooks
- Atomic Structure Test Questions eBook Subscription Services
- Atomic Structure Test Questions Budget-Friendly Options
- 6. Navigating Atomic Structure Test Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Atomic Structure Test Questions Compatibility with Devices
 - Atomic Structure Test Questions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Atomic Structure Test Questions
 - Highlighting and Note-Taking Atomic Structure Test Questions
 - Interactive Elements Atomic Structure Test Questions
- 8. Staying Engaged with Atomic Structure Test Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Structure Test Questions
- 9. Balancing eBooks and Physical Books Atomic Structure Test Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Structure Test Questions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Atomic Structure Test Questions
 - Setting Reading Goals Atomic Structure Test Questions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Atomic Structure Test Questions
 - Fact-Checking eBook Content of Atomic Structure Test Questions
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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