Name	Clans	Visite	
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Section: Concentration and Molarity

Complete each statement below by choosing a term from the following list. Use each term only once.

parts per million	moles	molarity	ratios	ister	
1	s the quantity o	f solute in a s	pecific quantity	y of solvent or	
solution. One way o	of expressing c	oncentration i	s to use the ur	ut of measure	
called	; it is the	ratio of		mount (in	
moles) to	volur	ne (in liters),	or simply	0	
solute per L of solu	tion. Table 1 is	your textbo	ok lists other	concentration	
units or	, such a	s	(m = r	nol solute/kg	
solvent) and	q)	pm = mL solv	ate/1 000 000.	0 mL solution).	

Solve the following problems and write your answer in the space provided.

2. What is the molarity of a hydrochloric acid solution that has a volume of 1500 mL and contains 441 g of HCl?

3. What is the molarity of a sugar solution that has a volume of 0.500 L and contains 17.1 g of ordinary sugar ($C_{12}H_{22}O_{11}$)?

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