

Name _____ Class _____ Date _____

Skills Worksheet

Concept Review**Section: Calculating Quantities in Reactions**

Complete each statement below by writing the correct term or phrase.

1. All stoichiometric calculations involving equations use _____ ratios.
2. When solving stoichiometric problems, you must _____ the equation first.
3. Balanced equations give the _____ numbers of moles of substances.
4. _____ in chemical equations provide mole ratios that can be used as conversion factors.
5. The conversion factor for converting between mass and moles is the _____ of the substance.
6. In making calculations involving _____, you must convert volume to mass.
7. To convert from volume to mass, you can use the _____ of the substance as the conversion factor.
8. When calculating the number of particles, you can use _____ as the conversion factor.

In the blanks at the left, write the letter of the choice that best completes the statement or answers the question. Consider the following problem when answering:

What mass of sulfuric acid is required to neutralize 2.65 g of potassium hydroxide? The products of the reaction are potassium sulfate and water.

- _____ 9. What should you do first after reading the problem carefully?
- a. Estimate the answer.
 - b. Calculate the molar mass of sulfuric acid.
 - c. Write a balanced chemical equation.
 - d. Convert all masses to moles.
- _____ 10. What should you do before setting up the problem?
- a. Determine the densities.
 - b. Calculate molar masses.
 - c. Convert all masses to moles.
 - d. Estimate the answer.

Concept Review Section 3 Chemistry

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Concept Review Section 3 Chemistry:

A Practical Guide to Scientific Writing in Chemistry Andrew Terhemen Tyowua, 2023-03-31 Successful completion of postgraduate studies especially PhD and career advancement in academia strongly depend on the ability to publish scientific papers or books and attract research grants However many chemical scientists find preparing scientific papers and research grant and book proposals difficult partly because of insufficient training in writing and partly because there are few practical books to enable them to learn the art This step by step practical guide is intended mainly for postgraduate students and early career researchers in chemical science and the libraries that serve them but will also be useful to other scientists

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Merrill Chemistry Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998

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