

Name _____

Class _____

Date _____

Skills Worksheet

Concept Review**Section: Concentration and Molarity**

Complete each statement below by choosing a term from the following list.
Use each term only once.

concentration

moles

solute

solution

liter

parts per million

molality

molarity

ratios

1. _____ is the quantity of solute in a specific quantity of solvent or solution. One way of expressing concentration is to use the unit of measure called _____; it is the ratio of _____ amount (in moles) to _____ volume (in liters), or simply _____ of solute per L of solution. Table 1 in your textbook lists other concentration units or _____, such as _____ ($m = \text{mol solute/kg solvent}$) and _____ ($\text{ppm} = \text{mL solute/1 000 000.0 mL solution}$).

Solve the following problems and write your answer in the space provided.

2. What is the molarity of a hydrochloric acid solution that has a volume of 1500 mL and contains 441 g of HCl?
3. What is the molarity of a sugar solution that has a volume of 0.500 L and contains 17.1 g of ordinary sugar ($\text{C}_{12}\text{H}_{22}\text{O}_{11}$)?

Concept Review Section Concentration And Molarity Answer Key

Shasha Hu



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