



Section 2.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

19
20
Hydrogen
Helium
Group 1

Atomic Mass	2.0
Atomic Number	2.0
Electron Configuration	1s ²
Chemical Name	Helium
Chemical Symbol	He

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 19. It exists as the periodic table.
- _____ 20. It has two the periodic table.
- _____ 21. Group 1 elements.
- _____ 22. Elements that are strong and conduct electricity.
- _____ 23. Group 2 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- _____ 24. There are two representative elements of elements.
- _____ 25. Most elements are located in the periodic table are metals.
- _____ 26. Group 1A elements except for hydrogen are located in the alkali metals.
- _____ 27. Group 2A elements are the alkaline earth metals.
- _____ 28. Group 3A elements are highly reactive transition elements in hydrogen.
- _____ 29. Group 4A elements are very reactive elements located in transition metals.
- _____ 30. Alkaloids have properties of both metals and non-metals.

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Y Pai



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