- Convert the following word equations into balanced chemical equations.
  - (a) copper metal(s) + chlorine(g) → copper(II) chloride(s)

    Cu(s) + cl<sub>2</sub>(g) → cu(l<sub>2</sub>
  - (b) manganese(II) carbonate(s) → manganese(II) oxide(s) + carbon dioxide(g)

    M9(C03) → M90 + C02
- (c) potassium metal(s) + water(1)  $\rightarrow$  potassium hydroxide(aq) + hydrogen(g)  $K(S) + H_2O(L) \longrightarrow KOH(qq) + H(G)$ (d) barium chloride(aq) + sodium sulfate(aq)  $\rightarrow$  barium sulfate(s) + sodium chloride(a  $BaCl_2(qq) + Na_2Sl_3(qq) \longrightarrow Ba(Sl_3(S) + Na_2Cl_2(qq) \longrightarrow Ba(Sl_3(S) + Na_2Cl_2(qq$
- (e) acetic acid(aq) + aluminum hydroxide(s)  $\rightarrow$  aluminum acetate(aq) + water(1)  $C_2H_3O_2$  + Al  $OH_3$ .  $\rightarrow$  Al  $(C_2H_3O_2)_3$  +  $H_2O$

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**Charles Henrickson** 

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