How to Balance a Chemical Equation

Reactants

 $N_2 + H_2$

Atom Counts

Nitrogen: 2

From the Subscript

Hydrogen: 2

Product
NH-

Atom Counts

Nitrogen: 1

Hydrogen: 3



When balancing a chemical equation, there should be equal numbers of every atom on both sides of the equation

 $N_2 + 3H_2 \longrightarrow 2NH_3$

Nitrogen: 2

Hydrogen: 3*2=6

From the Coefficient

Nitrogen: 2*1=2

Hydrogen: 2*3=6



Concept Review Answers Writing Balanced Equations

VM Jensen

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