

Concept Review

Section: Ionic and Covalent Bonding

1. Explain why atoms will often join together to form bonds.

2. Contrast ionic and covalent bonds.

3. Explain why a triple bond between two nitrogen atoms is stronger than a double bond between two oxygen atoms.

4. Explain how it is possible for a compound to have both ionic and covalent bonds.

5. Predict whether a gold ring would be a good conductor of electricity. What kind of bonds does gold have? How do these bonds explain gold's properties?

Concept Review Section Covalent Bonds 6 1 Answers

Rachel S Tattersall



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