

**2.1****PROPERTIES OF MATTER****Section Review****Objectives**

- Identify physical properties and physical changes
- Distinguish intensive properties from extensive properties
- Differentiate among three states of matter

**Vocabulary**

- |                      |                     |                   |
|----------------------|---------------------|-------------------|
| • mass               | • substance         | • gas             |
| • volume             | • physical property | • vapor           |
| • extensive property | • solid             | • physical change |
| • intensive property | • liquid            |                   |

**Part A Completion**

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

Properties used to describe matter can be classified as \_\_\_\_\_ 1 \_\_\_\_\_ or \_\_\_\_\_ 2 \_\_\_\_\_. The \_\_\_\_\_ 3 \_\_\_\_\_ of an object is a measure of the amount of matter the object contains. The \_\_\_\_\_ 4 \_\_\_\_\_ of an object is a measure of the space occupied by the object. An extensive property is one that depends on the \_\_\_\_\_ 5 \_\_\_\_\_ of matter. An intensive property is one that depends on the \_\_\_\_\_ 6 \_\_\_\_\_ of matter.

A \_\_\_\_\_ 7 \_\_\_\_\_ is matter that has uniform and definite composition. A solid has a definite \_\_\_\_\_ 8 \_\_\_\_\_ and \_\_\_\_\_ 9 \_\_\_\_\_. A liquid has a definite volume, but takes the \_\_\_\_\_ 10 \_\_\_\_\_ of its container. A \_\_\_\_\_ 11 \_\_\_\_\_ takes both the shape and volume of its container.

1. \_\_\_\_\_
2. \_\_\_\_\_
3. \_\_\_\_\_
4. \_\_\_\_\_
5. \_\_\_\_\_
6. \_\_\_\_\_
7. \_\_\_\_\_
8. \_\_\_\_\_
9. \_\_\_\_\_
10. \_\_\_\_\_
11. \_\_\_\_\_

**Part B True-False**

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- \_\_\_\_\_ 11. Matter has mass and occupies space.
- \_\_\_\_\_ 12. A liquid has a definite shape.
- \_\_\_\_\_ 13. Heating a solid to 200°C will cause it to change to a liquid.
- \_\_\_\_\_ 14. Gases are easier to compress than liquids.

# Concept Review Section Describing Matter Answers

**Susan G. Salvo**



## Concept Review Section Describing Matter Answers:

**Holt Chemistry** Holt Rinehart & Winston, 2003-01-24      **Matter** Prentice-Hall Staff, 1994      Concepts in Biology' 2007 Ed. 2007 Edition ,      **Matter, Building Block of the Universe** , 1993      **Human Biology** Daniel Chiras, 2012 Written for the introductory human biology course the Seventh Edition of Chiras acclaimed text maintains the original organizational theme of homeostasis presented in previous editions to present the fundamental concepts of mammalian biology and human structure and function Chiras discusses the scientific process in a thought provoking way that asks students to become deeper more critical thinkers The focus on health and homeostasis allows students to learn key concepts while also assessing their own health needs An updated and enhanced ancillary package includes numerous student and instructor tools to help students get the most out of their course      **Chemistry: Core Concepts, 3rd Edition** Allan Blackman, Daniel Southam, Gwendolyn Lawrie, Natalie Williamson, Christopher Thompson, 2024-01-09 The third edition of Chemistry Core Concepts Blackman et al has been developed by a group of leading chemistry educators for students entering university with little or no background in chemistry Available as a full colour printed textbook with an interactive eBook code this title enables every student to master concepts and succeed in assessment Lecturers are supported with an extensive and easy to use teaching and learning package      **Ebook: Chemistry: The Molecular Nature of Matter and Change** Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change      **States of Matter, States of Mind** Allan F.M. Barton, 2021-08-29 States of Matter States of Mind is an easy to read introduction to the way the physical world is put together and stays together The book presents the fundamental ideas and particles of the makeup of the universe to enable understanding of matter and why it behaves in the way it does Written in an engaging manner the book explains some of the intricate details and grand schemes of life and the universe by making analogies with common everyday examples For example the recipe for a cake tells us nothing of how good the cake tastes but is a model of the food and a scientific model is no closer to the reality of the materials than a recipe is to the mouth watering flavor of the cake Illustrated with helpful cartoons this book provides a vast knowledge of atoms and atmospheres The first several chapters introduce terms and fundamental ideas while later chapters deal successively with particles and systems from the electron to the universe as a system Each new idea introduced builds upon the last A user friendly bibliography provides references for further reading      Holt Chemistry Ralph Thomas Myers, 2004      **Matter and Energy** , 2005      **Applied Mechanics Reviews** , 1973      **Introductory Differential Equations** Martha L. Abell, James P. Braselton, 2023-12-21 2025 Textbook and Academic Authors Association TAA McGuffey Longevity Award Winner Introductory Differential Equations Sixth Edition provides the foundations to assist students in learning not only how to read and understand differential equations but also how to read technical material in more advanced texts as they progress through their studies The book's accessible explanations and many robust sample problems are appropriate for a first semester course in introductory ordinary differential equations

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