

Name _____

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FLAME TEST LAB



Every atom consists of a nucleus with tiny electrons whizzing round it. The further away from the nucleus they are, the more energy the electrons have. If a metal atom is heated, the electrons get enough energy to jump higher away from the nucleus. When they fall back closer to the nucleus, they give off this extra energy as light.

Different metals produce different coloured light. If we look at the colour of the light made when a solution of metal is heated in a flame, we can tell which metal is there.

PRE-LAB DISCUSSION:

All metal elements have distinct properties. One of these properties is its own special color when burned in an open flame. This is caused by the flame exciting the electrons of the metal element in a compound. When the electrons return from the excited state to the ground state, a photon of energy is released. Some of the energy is in our visible light spectrum, and therefore we can observe the different colors of the different metals. (The finger prints of metals) Only the metal in the compound is affected by the flame.

PURPOSE:

In this experiment, we shall use a flame to give atoms energy, and observe the flame color and bright-line spectrum. We will then use this data to identify an unknown atom. You will learn to identify some metals by their special color. We will investigate barium, calcium, copper, lead, potassium and sodium, as they give readily identifiable colors. You must work very carefully so your chemicals remain pure.

PROCEDURE:

1. Carefully light your Bunsen burner. Adjust the air intake collar so the flame is blue in the center.
2. Obtain a set of wooden splints and the chemicals to be tested.
3. Dip the splint into the glass vial containing the compound your teacher has provided.
4. Be Sure to hold splint by the bottom. Place one end into the flame to see the color that it makes. Be careful not to let the chemicals fall into the flame. Tip the burner on an angle (*do not hold the splint directly over the flame*) and place splint in the colorless part of the flame.
5. Observe the color of the flame produced when the compound burns. Record the name of the metal and the color of its flame in the Data Table provided.

Conclusion Of Flame Test Lab

Kellie Marcarelli



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Practical Chemistry Labs Leonard Saland, 1989 Features self contained step by step activities using common materials and covering topics from food chemistry to papermaking and electrochemistry Illustrates the connection between the real world and chemistry concepts such as solutions chemistry acids and bases and more Includes teacher notes quizzes and answers to help monitor student progress

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Alcamo's Laboratory Fundamentals of Microbiology Jeffrey C. Pommerville, 2007 Microbiology

Technical Report Cold Regions Research and Engineering Laboratory (U.S.), 1976

Teaching Science With Interactive Notebooks Kellie Marcarelli, 2010-05-18

Packed with student samples this resource describes how to implement interactive notebooks in the inquiry based science classroom including execution time management and grading

Hard Bound Lab Manual Biology Neena Sinha, R Rangarajan, R P Manchanda, R K Gupta, Rajesh Kumar, Lab Manuals

Gourmet Lab Sarah Reeves Young, 2011 Hands on inquiry based and relevant to every student OCOs life Gourmet Lab serves up a full menu of activities for science teachers of grades 6 OCo12 This collection of 15 hands on experiments OCOeach of which includes a full set of both student and teacher pages OCOchallenges students to take on the role of scientist and chef as they boil bake and toast their way to better understanding of science concepts from chemistry biology and physics By cooking edible items such as pancakes and butterscotch students have the opportunity to learn about physical changes in states of matter acids and bases biochemistry and molecular structure The Teacher pages include Standards addressed in each lab a vocabulary list safety protocols materials required procedures data analysis student questions answer key and conclusions and connections to spur wrap up class discussions Cross curricular notes are also included to highlight the lesson OCOs connection to subjects such as math and literacy Finally optional extensions for both middle school and high school levels detail how to explore each concept further What better topic than food to engage students to explore science in the natural world

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benefit from the diverse perspectives and innovative approaches presented in these papers inspiring new ideas and solutions for their own research endeavors Moreover the positive influence of this conference extends beyond the current discussions It is expected that the findings and recommendations shared in these proceedings will serve as a foundation for future research in the field of economic management and modeling engineering By fostering collaboration knowledge sharing and academic discourse this conference aims to contribute to the advancement of the field and stimulate further research initiatives in the years to come

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Laboratory Manual for Principles of General Chemistry J. A. Beran, Mark Lassiter, 2022-08-16 Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure selecting the proper apparatus employing the proper techniques systematically analyzing and interpreting the data and minimizing inherent variables As a result of good data a scientific and analytical conclusion is made which may or may not be right but is certainly consistent with the data Experiments write textbooks textbooks don't write experiments A student's scientific literacy grows when experiences and observations associated with the scientific method are encountered Further experimentation provides additional cause effect observations leading to an even better understanding of the experiment The 11th edition's experiments are informative and challenging while offering a solid foundation for technique safety and experimental procedure The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit An Additional Notes column is included in each

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