

## Balancing Equations Worksheet

- 1) \_\_\_\_  $\text{Na}_3\text{PO}_4$  + \_\_\_\_  $\text{KOH}$   $\rightarrow$  \_\_\_\_  $\text{NaOH}$  + \_\_\_\_  $\text{K}_3\text{PO}_4$
- 2) \_\_\_\_  $\text{MgF}_2$  + \_\_\_\_  $\text{Li}_2\text{CO}_3$   $\rightarrow$  \_\_\_\_  $\text{MgCO}_3$  + \_\_\_\_  $\text{LiF}$
- 3) \_\_\_\_  $\text{P}_4$  + \_\_\_\_  $\text{O}_2$   $\rightarrow$  \_\_\_\_  $\text{P}_2\text{O}_3$
- 4) \_\_\_\_  $\text{RbNO}_3$  + \_\_\_\_  $\text{BeF}_2$   $\rightarrow$  \_\_\_\_  $\text{Be}(\text{NO}_3)_2$  + \_\_\_\_  $\text{RbF}$
- 5) \_\_\_\_  $\text{AgNO}_3$  + \_\_\_\_  $\text{Cu}$   $\rightarrow$  \_\_\_\_  $\text{Cu}(\text{NO}_3)_2$  + \_\_\_\_  $\text{Ag}$
- 6) \_\_\_\_  $\text{CF}_4$  + \_\_\_\_  $\text{Br}_2$   $\rightarrow$  \_\_\_\_  $\text{CBr}_4$  + \_\_\_\_  $\text{F}_2$
- 7) \_\_\_\_  $\text{HCN}$  + \_\_\_\_  $\text{CuSO}_4$   $\rightarrow$  \_\_\_\_  $\text{H}_2\text{SO}_4$  + \_\_\_\_  $\text{Cu}(\text{CN})_2$
- 8) \_\_\_\_  $\text{GaF}_3$  + \_\_\_\_  $\text{Cs}$   $\rightarrow$  \_\_\_\_  $\text{CsF}$  + \_\_\_\_  $\text{Ga}$
- 9) \_\_\_\_  $\text{BaS}$  + \_\_\_\_  $\text{PtF}_2$   $\rightarrow$  \_\_\_\_  $\text{BaF}_2$  + \_\_\_\_  $\text{PtS}$
- 10) \_\_\_\_  $\text{N}_2$  + \_\_\_\_  $\text{H}_2$   $\rightarrow$  \_\_\_\_  $\text{NH}_3$
- 11) \_\_\_\_  $\text{NaF}$  + \_\_\_\_  $\text{Br}_2$   $\rightarrow$  \_\_\_\_  $\text{NaBr}$  + \_\_\_\_  $\text{F}_2$
- 12) \_\_\_\_  $\text{Pb}(\text{OH})_2$  + \_\_\_\_  $\text{HCl}$   $\rightarrow$  \_\_\_\_  $\text{H}_2\text{O}$  + \_\_\_\_  $\text{PbCl}_2$
- 13) \_\_\_\_  $\text{AlBr}_3$  + \_\_\_\_  $\text{K}_2\text{SO}_4$   $\rightarrow$  \_\_\_\_  $\text{KBr}$  + \_\_\_\_  $\text{Al}_2(\text{SO}_4)_3$
- 14) \_\_\_\_  $\text{CH}_4$  + \_\_\_\_  $\text{O}_2$   $\rightarrow$  \_\_\_\_  $\text{CO}_2$  + \_\_\_\_  $\text{H}_2\text{O}$
- 15) \_\_\_\_  $\text{Na}_3\text{PO}_4$  + \_\_\_\_  $\text{CaCl}_2$   $\rightarrow$  \_\_\_\_  $\text{NaCl}$  + \_\_\_\_  $\text{Ca}_3(\text{PO}_4)_2$
- 16) \_\_\_\_  $\text{K}$  + \_\_\_\_  $\text{Cl}_2$   $\rightarrow$  \_\_\_\_  $\text{KCl}$
- 17) \_\_\_\_  $\text{Al}$  + \_\_\_\_  $\text{HCl}$   $\rightarrow$  \_\_\_\_  $\text{H}_2$  + \_\_\_\_  $\text{AlCl}_3$
- 18) \_\_\_\_  $\text{N}_2$  + \_\_\_\_  $\text{F}_2$   $\rightarrow$  \_\_\_\_  $\text{NF}_3$
- 19) \_\_\_\_  $\text{SO}_2$  + \_\_\_\_  $\text{Li}_2\text{Se}$   $\rightarrow$  \_\_\_\_  $\text{SSe}_2$  + \_\_\_\_  $\text{Li}_2\text{O}$
- 20) \_\_\_\_  $\text{NH}_3$  + \_\_\_\_  $\text{H}_2\text{SO}_4$   $\rightarrow$  \_\_\_\_  $(\text{NH}_4)_2\text{SO}_4$

# Balancing Equations Practice With Answers

**Julia Burdge**



## Balancing Equations Practice With Answers:

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web tone deafness is also associated with other musical specific impairments such as the inability to keep time with music beat deafness or the lack of rhythm or the inability to remember or recognize a song these disabilities can appear separately but some research shows that they are more likely to appear in tone deaf people

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