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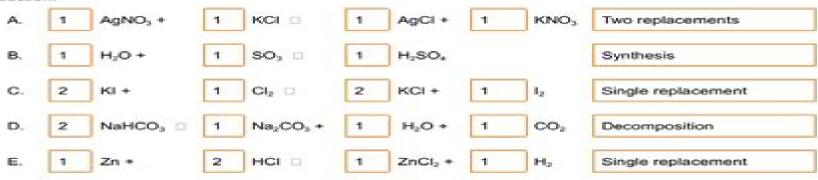
Activity B:	Get the Gizmo ready:	1
Classifying reactions	Turn off Show summary and Show histograms.	5° 9

Introduction: Chemical equations show how compounds and elements react with one another. An element is a substance consisting of one kind of atom, such as aluminum (Al) or oxygen gas (O_2) . A compound is a substance made of more than one kind of atom, such as water (H_2O) or table salt (NaCl).

Question: How are chemical reactions classified?

- Match: Most chemical reactions can be classified as one of four types. Using the chemical equations in the Gizmo as a guide, match the following definitions to the type of reaction.
 - One reactant is broken down into two or more products. 8 A. Synthesis A fuel combines with oxygen to produce carbon dioxide (CO₃) and Decomposition water (H₂O). Two or more reactants combine to form one product. C. Single replacement pa, D) Two compounds react to form two different compounds. D. Double replacement C A compound reacts with an element to form a new compound and a E. Combustion different element.

 Practice: Balance each of the chemical equations below. (Some equations may already be in balance.) In the space to the right, classify the reaction as a synthesis, decomposition, single replacement, double replacement, or combustion.



Balancing Equations Gizmo Key

LL Leslie

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